

Catalyst Performance Comparison		
Reaction Conditions and Performance	Our Data (Table 5)	Literature data*
GHSV, h ⁻¹	3750	180
T, ° C.	280	200
P, MPa	5.4 MPa	0.1
Catalyst	Rh—Mn/SiO ₂	Rh ₂ MnO ₄ /SiO ₂
CO Conversion, %	24.6	20.2
Selectivity		
CH ₄	38.4	42.3
CO ₂	0	3.0
MeOH	3.9	2.0
EtOH and C ₂ Oxygenates	56.1	20.4
Other HCs	1.6	32.3

*S. Ishiguro, S. Ito, K. Kunimori, *Catalysis Today* 45, 197-201, 1998 (Table 1)

As can be seen, the use of a microchannel reactor allows us to operate at high throughput to achieve high conversion and improved selectivity.

TABLE 6

Performance Comparison of Structured Rh—Mn/SiO ₂ Catalyst with Identical Powdered Form in a MicroChannel Reactor		
	Run numbers	
	Run EC-02	Run ET 32
Catalyst Configuration	Rh—Mn/SiO ₂ coated on FeCrAlY felt	Powdered Rh—Mn/SiO ₂
GHSV, h ⁻¹	20,000	2700
Conversion, mol %	20.4	22.7
Selectivity, %		
CH ₄	36.5	31.1
CO ₂	2.3	4.7
C ₂ ⁺ HCS	3.2	1.7
Alcohols and C ₂ ⁺ Oxy	58.0	62.4
Specific Activity, mmolCO Converted/g · h	46.0	26.8

H₂/CO = 1:1, T = 300° C.

We claim:

1. A method of synthesizing alcohols from CO or CO₂ comprising:

flowing a reactant gas mixture comprising H₂ and CO or CO₂ into contact with a catalyst;
wherein the catalyst comprises a Pd—Zn alloy dispersed on alumina; and
forming an alcohol or alcohols.

2. The method of claim 1 wherein the alcohol or alcohols formed in the step of forming an alcohol or alcohols consists essentially of methanol.

3. The method of claim 1 wherein the catalyst further comprises a Fisher-Tropsch catalyst and wherein the alcohol or alcohols formed in the step of forming an alcohol or alcohols comprises higher alcohols that contain 2 or more carbon atoms.

4. The method of claim 3 wherein the alcohol or alcohols formed in the step of forming an alcohol or alcohols comprises a mixture of alcohols in which ethanol is the principle alcohol.

5. The method of claim 3 wherein the catalyst comprises the Pd—Zn alloy dispersed on alumina catalyst and a Fisher-Tropsch catalyst that are mixed together.

6. The method of claim 3 wherein the catalyst comprises a first section that consists essentially of the Pd—Zn alloy dispersed on alumina catalyst, and a second section that comprises the Fisher-Tropsch catalyst.

7. The method of claim 3 wherein said step of flowing is controlled so that the contact time is less than 1 second.

8. The method of claim 7 wherein the catalyst is disposed in a reaction channel having a width of 5 mm or less, and further wherein the temperature variation across the catalyst is 10° C. or less.

9. The method of claim 6 wherein the first section and the second section are disposed in a reaction channel having a width of 5 mm or less.

10. The method of claim 6 wherein the reactant gas mixture contacts the first section before contacting the second section.

11. The method of claim 1 wherein the catalyst comprises crystalline ZnO.

12. The method of claim 2 wherein the reactant gas mixture comprises CO, and CO reacts with H₂ to form methanol.

13. The method of claim 1 wherein the reactant gas mixture comprises CO and CO₂.

14. The method of claim 1 wherein the reactant gas mixture consists essentially of CO and H₂.

15. A method of synthesizing ethanol or higher alcohols from CO₂ comprising:

flowing a reactant gas mixture comprising CO₂ and H₂ into contact with a catalyst;

wherein the catalyst comprises: (a) Pd—Zn alloy dispersed on alumina and (b) a Fischer-Tropsch catalyst; and
forming ethanol or higher alcohols.

16. A method of synthesizing an alcohol comprising: contacting hydrogen and CO over an alcohol catalyst in a microchannel;

removing heat into a heat exchanger; and
converting at least 20% of the CO into products with a selectivity to C₂⁺ oxygenates of at least 30%.

17. The method of claim 16 comprising a selectivity to ethanol of at least 30%.

18. The method of claim 16 wherein the catalyst is disposed in a flow-by configuration in the microchannel.

19. The method of claim 16 wherein the catalyst comprises an alcohol synthesis catalyst and a Fischer-Tropsch catalyst.

20. The method of claim 19 wherein the alcohol synthesis catalyst and Fischer-Tropsch catalyst are mixed together.

21. The method of claim 19 wherein the alcohol synthesis catalyst and Fischer-Tropsch catalyst are sequentially arranged in the microchannel.

22. The method of claim 16 wherein the catalyst comprises Rh and Mn disposed on silica, titania, or zirconia.

23. The method of claim 22 wherein the catalyst is disposed on a large pore support and wherein the catalyst on the large pore support has a pore volume in which at least 20% of the pore volume is composed of pores in the size range of 0.1 to 300 microns.

24. The method of claim 16 wherein temperature is maintained at less than about 270° C.

25. The method of claim 22 wherein temperature is maintained at less than about 270° C.

26. The method of claim 16 wherein and comprising passing reactants through the microchannel at a gas hourly space velocity of at least 3000 h⁻¹ and converting at least 20% of the CO into products with a selectivity to C₂⁺ oxygenates of at least 40%.

27. The method of claim 18 wherein and comprising passing reactants through the microchannel at a gas hourly space velocity of at least 3000 h⁻¹.